

## RED WINE PRODUCTION CONSIDERATIONS

*Learning Outcomes: The reader will understand the production parameters that impact stylistic red wine production.*

### Chapter Outline

**A Review of Tannins and Anthocyanins**

**Maceration during Red Winemaking**

**Grape Maturity**

**Crushing/Destemming**

**Stems**

**Tank Type and Size**

**Open versus Closed Tanks**

**Cap Management**

**Maceration Period**

**Controlled Aeration**

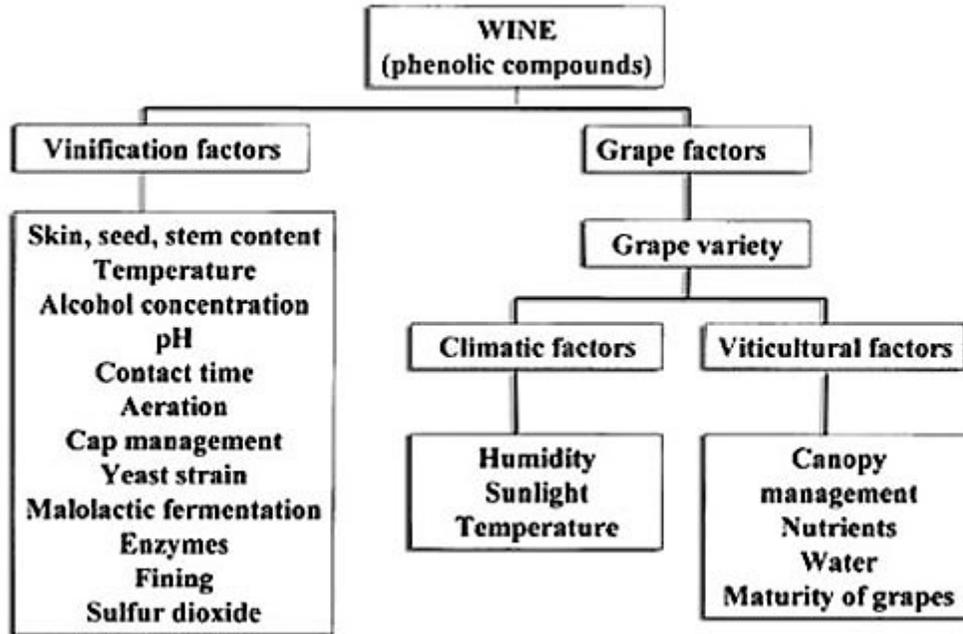
**Red Wine Production Considerations for Virginia**

### Section 1.

#### **A Review of Tannins and Anthocyanins**

The phenolic compounds in wines are the result of a host of factors, including those depicted in Figure 1.

**Figure 1. Factors Affecting Phenolic Compounds in Wines** (in part, from Stockley and Høj, 2005)



Approximately one-third of the carbon produced by grapevines is used to make phenolic substances. Therefore, phenols are an important constituent of both grapes and wines. It is the structure of a wine, not the composition that determines flavor and integration. The integration of the structural and textural elements, and has been discussed in editions of *Enology Notes* (see [Enology Notes #52, 68, 69, 76, 84, 107, 108, and 113](#)).

Kassas and Kennedy (2011) noted some interesting positive correlations in red wines. In their study, wines commanding the highest price in the marketplace had several attributes:

- highest concentrations of total tannins
- highest concentration of skin tannins
- degree of phenol polymerization

The term *tannin* defines a very heterogeneous group of phenolic compounds that are identified, based on certain properties:

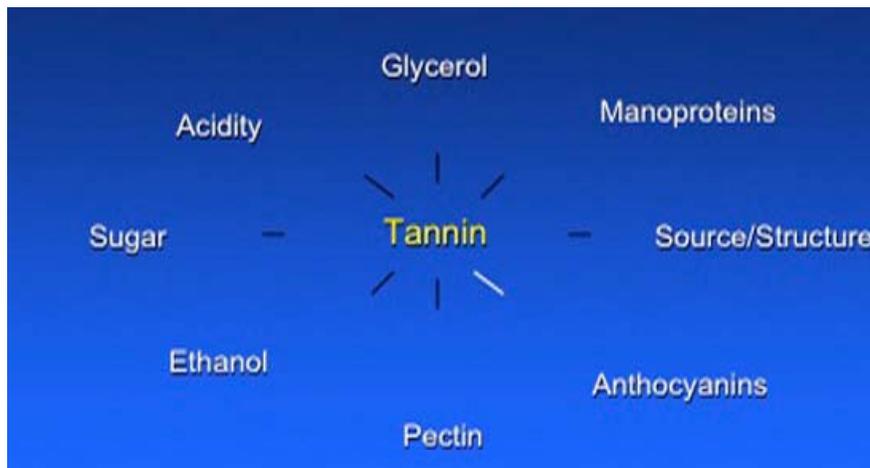
- astringency
- bitterness
- reaction with ferric chloride
- ability to bind with proteins, e.g., tannin leather – hence, the term *tannins*

It was their characteristic interaction with proteins that traditionally differentiated tannins from other phenols. However, not all phenols that bind with proteins elicit an astringent response, and tannins are not the only compounds in wines that cause astringency.

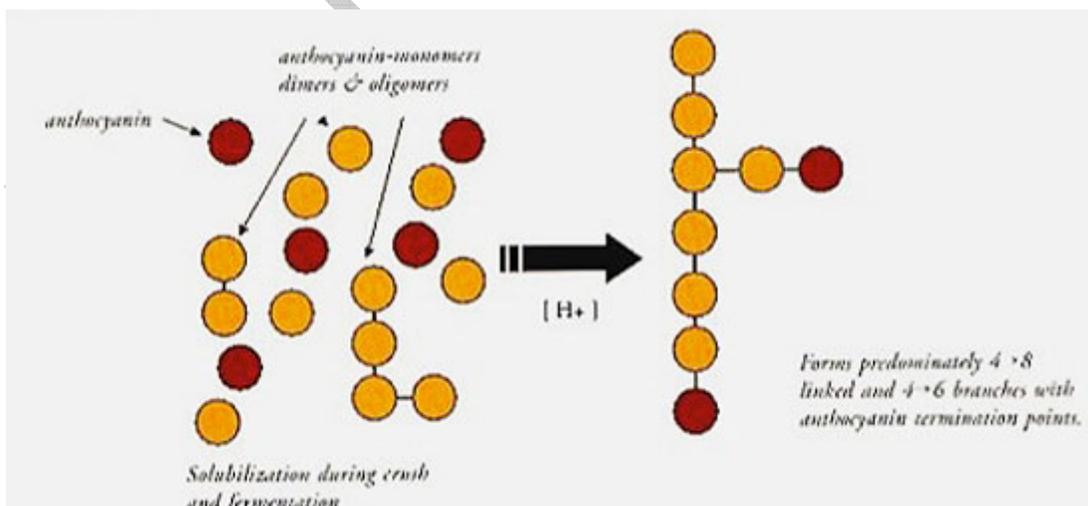
Some phenols, including tannins, have the ability to polymerize, or associate, with themselves and other compounds, including anthocyanin pigments. As polymerization occurs, the molecule becomes larger. The number of subunits bound together is referred to as the DP number, or degree of polymerization. So-called tannin “quality” relates to the following:

- degree of polymerization
- association of tannins with other molecules
- stereospecific nature of the molecule, which can make it harsh and hard, or supple

Grape tannins derived from the skin, seeds, and stems differ in their sensory properties, and their DP number and overall subunit concentration. Tannin perception is a function of the factors shown in Figure 2.

**Figure 2. Factors Influencing Tannin Perception****Pigment-Tannin Polymerization**

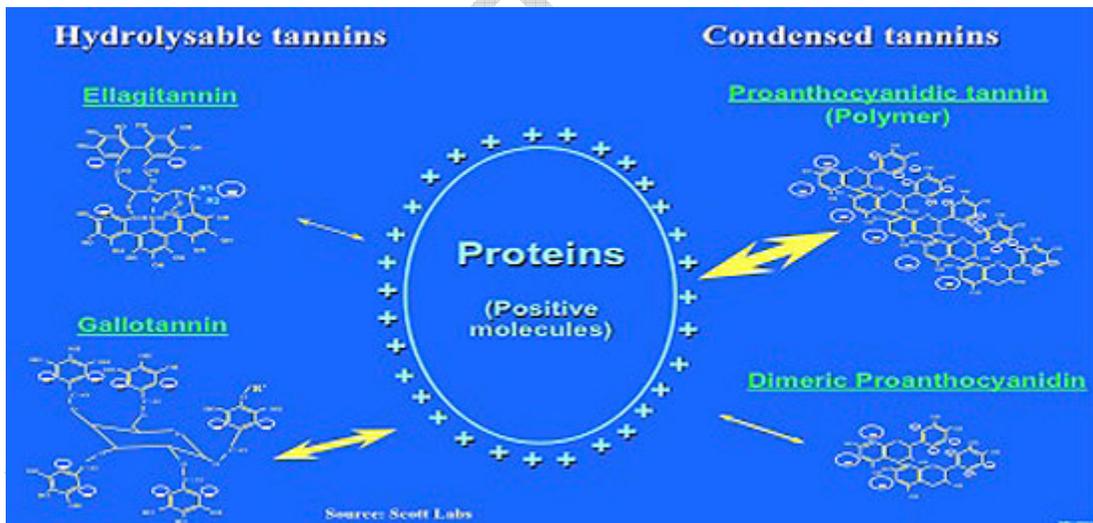
In grapes and wines, anthocyanin pigments can be either free monomers, that is, unbound, or associated with other phenols to form polymers.

**Figure 3. Proposed Condensation Reactions under Non-Oxidative Conditions (McCord, 1990)**

Anthocyanin-tannin polymerization occurs both, in the fruit during maturation, and in wines. Polymerization continues until an anthocyanin molecule binds the terminal end of the tannin chain, thus stopping the polymerization (see Figure 3).

As such, the ratio of anthocyanins to tannins is important in impacting the extent of polymerization. This is highly crucial, because the degree of polymerization affects astringency, an important feature for Virginia red wines. Large tannin polymers provide a relatively large number of binding sites to interact with proteins, including salivary proteins (Figure 4). Wines with an abundance of large polymers tend to lack softness, may lack color stability, and often possess a dry mouth sensation.

**Figure 4. Reactivity of Tannins** (Scott Labs)



Smaller polymers, on the other hand, have fewer protein binding sites. As such, they produce less astringency, and provide a greater degree of soft tannins and more palate depth. Additionally, these smaller pigment polymers provide a greater reductive strength (see below).

The more anthocyanins, the shorter the resulting polymers and the finer the tannins. Smaller polymers lead to smaller colloids which have a softer mouthfeel (Smith 2010).

Binding with phenols not only can involve other phenols like anthocyanins but also hydrogen sulfide, mercaptans and sulfur containing amino acids (Smith 2011).

### Reductive Strength

Longevity, or the ability to age, is an important quality attribute. The reductive strength of a wine is a measure of the uptake of oxygen. This is influenced by the phenol composition and lees, among other things.

Some phenols, including tannins, have the ability to react with oxygen, bind with another phenol and recreate the original structure-thus allowing it to react over and over again.

Remarkably, the reaction of a young wine with oxygen can make that wine more resistant to later oxidation. This means young wines can consume oxygen and this actually increases reductive strength.

The problems with under-ripe fruit include the following (Smith 2011):

- insufficient pigments
- limited extraction
- limited desirable flavors, which limits tannin capacity

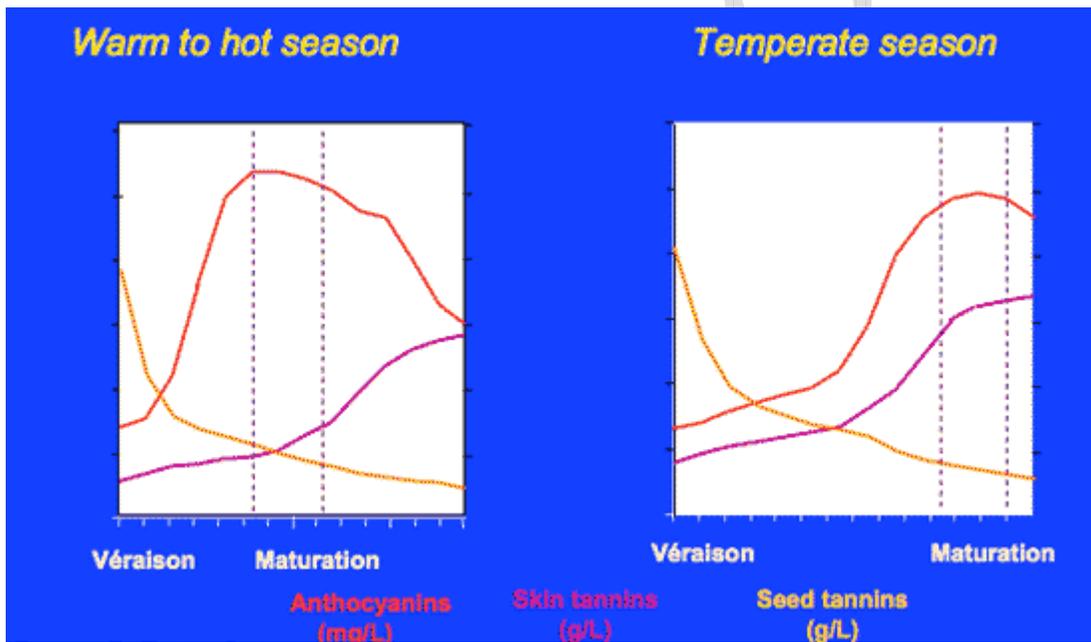
The problems with over-ripe fruit include the following (Smith 2011):

- loss of color

- high alcohol capacity, which can destabilize color
- significant loss of reductive strength in the resultant wine

The change in the phenolic content, as a function of excessive fruit maturity, can lower the reductive strength by a factor of 10 (Smith, 2011). The following (Figure 5) illustrates the variation that can occur between the skin tannins and anthocyanin concentration among seasons. While seed tannins are less affected, note the large potential difference in the ratio of tannins to anthocyanins.

**Figure 5. Effect of climate on relative phenolic concentration (Lallemande)**



Grapes producing the most intense, balanced wines with the greatest longevity usually have a high anthocyanin to tannin ratio. Indeed, wine quality may be dependent, in part, on this ratio (see [Enology Notes #8, 103](#)). While longevity has not been a traditional quality parameter in Virginia, it should be. In order for the industry to be taken seriously our red wines should improve with time, not simply survive.

### Factors Impacting Red Wine Color

Color is an important wine attribute, because humans are visually oriented. As such, wine color can certainly bias evaluations. A classic example of color bias is to change the color of a white wine, such as Chardonnay, with red food coloring. In blind evaluations, the color-adjusted wine frequently receives a different sensory rating for attributes such as fullness, body, and complexity.

As such, richly-colored wines are assumed to have high volume or body, and softer tannins. Conversely, a wine with less color is automatically assumed to have “green” or “harsh” tannins.

Spectral color in wine is a function of these three elements:

- anthocyanin concentration
- polymeric pigments
- concentration of cofactors, or certain non-colored compounds, which bind with anthocyanins

Hyperchromicity, also known as *copigmentation*, is an interesting phenomenon that allows more visible red color than would be expected due to the anthocyanin concentration alone. Cofactors are non-colored compounds that have the ability to bind with anthocyanins, creating more color than the unbound pigment, hence the term *hyperchromicity*.

The concentration and type of cofactors vary greatly from variety to variety, season to season, but include some non-flavonoid phenols, flavonols, and the amino acid arginine. It is not likely that enological tannins contain compounds that act as cofactors.

Because red color is a function of three elements (anthocyanin concentration, cofactor concentration, and polymeric pigments), it is possible to have the following (Boulton, 2005):

- Change in grape anthocyanin concentration = Change in wine color
- Change in grape anthocyanin  $\neq$  Change in wine color
- No change in grape anthocyanin = Change in wine color

The above highlights several points:

- Variation in cofactors and polymeric pigment concentration may be more important to spectral color than simple anthocyanin concentration.
- Grape pricing based on anthocyanin concentration alone may not be desirable.
- Harvesting based on anthocyanin concentration will not necessarily assure desirable red wine color.

Young wines have a high concentration of unbound pigments. Once they are incorporated into polymers with tannins they are stabilized. Early stabilization via oxygen exposure is a significant benefit to color, texture, and aromatic properties. Because much of the phenol polymerization is oxidative anything that depletes the oxygen content early may be deferential. Lees contact can lower the oxygen content and therefore may reduce incorporation of anthocyanins into stable pigment polymers. This is a very important red wine production consideration for Virginia. Many store red wines on secondary lees to help provide mannoproteins for structural development. If that storage occurs prior to adequate pigment-tannin polymerization, the extent of polymerization could be reduced.